

## PHYTOCHEMICAL SCREENING, ANTIOXIDANT AND ANTIBACTERIAL ACTIVITIES OF EXTRACTS AND FRACTIONS OF *Dillenia suffruticosa* LEAVES

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### ABSTRACT

*Dillenia suffruticosa* or ‘Simpur bini’ is known to have ethnomedicinal properties and had been used traditionally to heal wounds, relieve fever and rheumatism. There has been limited studies carried out on this species, therefore, this study aims to evaluate the phytochemical contents, antioxidant and antibacterial activities of aqueous extract, methanol extract and its fractions obtained from the leaves of *D. suffruticosa*. The dried leaves were extracted using methanol before successive solvent partitioning was carried out on the extract using hexane, chloroform, ethyl acetate and diethyl ether. In addition to this, aqueous decoction was conducted. The antioxidant activities were determined using Total Phenolic Content (TPC), Total Flavonoid Content (TFC) and 2,2-diphenyl-1-picryl-hydrazyl-hydrate (DPPH) free radical scavenging methods. Phytochemical screening had shown that most of the extracts and fractions contained alkaloids, steroids, phenolics, flavonoids and saponins. The diethyl ether and ethyl acetate fractions showed higher TPC and TFC values. The diethyl ether and ethyl acetate fractions also showed higher antioxidant activities determined via DPPH assay. Antibacterial activities determined using disc diffusion assay showed the methanol extract and its fractions had antibacterial activity against *Staphylococcus aureus*, with the diethyl ether fraction having comparable activity with the standard antibiotic streptomycin. However, inhibition against *Bacillus subtilis* was only observed in hexane, chloroform and diethyl ether fractions. No inhibition was observed against *Escherichia coli* and *Pseudomonas aeruginosa*. This study identified the diethyl ether and ethyl acetate fractions of *D. suffruticosa* leaves as potential sources of bioactive compounds. Further investigations on the bioassay-guided isolation of bioactive compounds of this species may lead to the discovery of new pharmaceutical or effective antimicrobial agents.

**Key words:** *Dillenia suffruticosa*, Dilleniaceae, antioxidant activity, antibacterial activity, flavonoids, phenolics, phytochemical screening

### INTRODUCTION

The use of conventional medicine which is primarily based on plants play a vital role in health care applications, whereby about 80% of the world’s population are dependent on it (Tuama & Mohammed, 2018). *Dillenia suffruticosa* (Griff)

Martelli also locally known as ‘Simpur bini’, is an evergreen shrub which belongs to the Dilleniaceae family found in forests of Brunei Darussalam. The plant grows up to 7 metres high and has large spirally arranged leaves with the size of 37 x 25 cm. The flowers are scentless (about 10-13 cm wide) with bright yellow petals and white stamens that could bloom in less than a day and the fruits are bright pink with red seeds.

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*D. suffruticosa* has traditionally been used as a wound healer and to relieve fever (Goh *et al.*, 2017). Moreover, this plant has been used to treat rheumatism (Hanum, 1999) and as an astringent (Wiar *et al.*, 2004). Its twigs and leaf stalks can be broken off and the exudate applied to an external wound, while the young pounded leaves are used as poultice to stop bleeding (Department of Agriculture, 2000). A poultice of the leaves can also be applied to affected areas to relieve from inflammation. *D. suffruticosa* has also been used in postpartum treatment (Wiar *et al.*, 2004) and their fruits have been used in the treatment of cancerous growth (Graham *et al.*, 2000).

There is limited scientific literature that supports the ethnomedicinal use of *D. suffruticosa*. To date, only two studies (Wiar *et al.*, 2004; Armania *et al.*, 2013) have reported the bioactivities of this plant, using the methanol and aqueous extracts obtained from various parts of the plant such as roots, leaves, fruits and flowers. It was reported that the methanolic extracts from the roots had the best antioxidant activity compared to the other extracts, with the IC<sub>50</sub> value of 31.33 ± 1.15 µg mL<sup>-1</sup> (Armania *et al.*, 2013). Antimicrobial activity of methanolic extract from leaves of *D. suffruticosa* was also evaluated using the disc diffusion method against *Bacillus cereus*, *B. subtilis*, *Candida albicans* and *Pseudomonas aeruginosa* (Wiar *et al.*, 2004). This indicates that with fractionation and isolation of *D. suffruticosa*, more potent antioxidant and antibacterial activities can be further analysed.

The aim of the present study was to evaluate the extracts and fractions obtained from the leaves of *D. suffruticosa* for their phytochemical contents and for their antioxidant and antibacterial activities.

## MATERIALS AND METHODS

### Materials

The fresh leaves of *D. suffruticosa* were collected from Kampung Batu Ampar, Brunei Darussalam, in April 2014. A voucher specimen was deposited in the Universiti Brunei Darussalam Biology Herbarium under the reference number UBDH/FHY01.

All solvents (methanol, hexane, chloroform, ethyl acetate and diethyl ether), nutrient broth and Mueller Hinton agar (MHA) were purchased from Merck, Germany. All other chemicals were purchased from Sigma-Aldrich, Germany. Spectrophotometric measurements were carried out using 2 mL plastic cuvettes and a single beam UV spectrophotometer (Optizen 1412V, Mecasys, Korea).

### Preparation of extracts

The leaves were air-dried and subsequently powdered using an electric blender. A total of 3 kg of powdered leaves were extracted with 5 L methanol using Soxhlet extraction. The resulting solution was filtered using Whatman No. 2 filter paper and concentrated under vacuum and 654 g of methanolic crude extract was obtained. For further analysis, 50 g of this extract was stored at 4°C but the remainder was dissolved in 1 L of methanol and 100 mL of distilled water, followed by successive fractionation with 1 L of hexane, 1 L of chloroform and 1 L of ethyl acetate (Lakache *et al.*, 2016). After this successive fractionation, it was found that a small amount of orange-yellow precipitate was present in the remaining methanol-aqueous mixture and was found to be soluble in diethyl ether. Thus, a final step of fractionation was carried out using a total of 500 ml diethyl ether. All solvents were removed under vacuum using an IKA rotary evaporator to dryness, and the resulting solid residues were stored at 4°C until further use.

The aqueous extract was obtained by replicating the traditional methods used by the locals to prepare the *D. suffruticosa* decoction, which was by heating 5 g of the leaves in distilled water at 80°C for an hour. It was concentrated under vacuum and lyophilised using a freeze-drier overnight. The resulting powder was stored in a desiccator until further use. The systematic diagram of the preparation of the extracts is shown in Figure 1.

### Phytochemical screening

The extracts and fractions were each analysed for the presence of alkaloids, steroids, phenolics, flavonoids and saponins according to the methods previously described (Edeoga *et al.*, 2005; Ayoola *et al.*, 2008; Lakache *et al.*, 2016).

#### Alkaloids

0.2 g of the extract or fraction was treated with warm 2% sulphuric acid (5 mL) and filtered. A few drops of Dragendorff's reagent was added to the filtrate. Orange-red precipitate indicates the presence of alkaloids.

#### Steroids

0.5 g of the extract or fraction was treated with chloroform (5 mL) and filtered. A few drops of concentrated sulphuric acid were carefully added into the filtrate, shaken and allowed to stand. The formation of red or yellow precipitate indicates the presence of steroids.

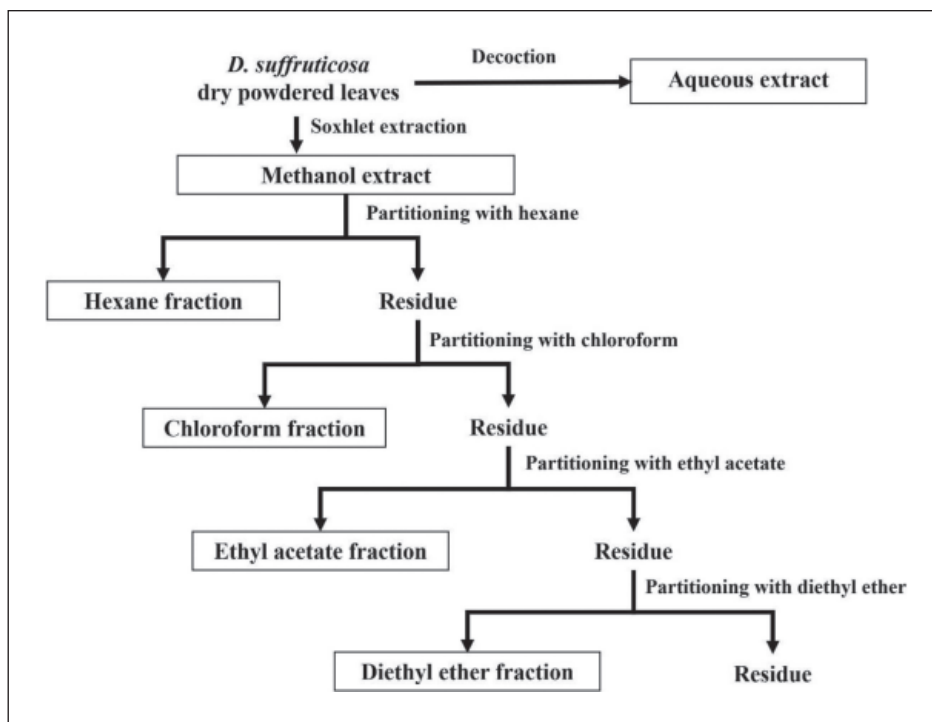


Fig. 1. The systematic diagram of the preparation of extracts.

#### Phenolic compounds

3 to 5 drops of 10% aqueous iron (III) chloride were added to 1 mL of the extract or fraction. The presence of phenolic compounds was detected by a change in colour from the initial bright yellow solution to form a dark green solution.

#### Flavonoids

5 mL of 10% sodium hydroxide was added to 2 mL of the extract or fraction. The formation of an intense yellow solution, which decolourised upon the addition of a few drops of dilute acid, indicates the presence of flavonoids.

#### Saponins

The extract or fraction (1 g) was diluted with distilled water in a 15 mL centrifuge tube and shaken vigorously. The formation of 1 cm layer of foam that persists for 15 minutes indicates the presence of saponins.

#### Antioxidant activities

##### Total phenolic content (TPC)

The TPC was estimated using the Folin-Ciocalteu's assay (Ebrahimzadeh *et al.*, 2008). To each of the 0.5 mL aliquots of various concentrations of gallic acid (1, 5, 10, 25, 50, 100, 250 and 500 mg mL<sup>-1</sup>, dissolved in methanol) or 1000 µg mL<sup>-1</sup> of the extract or fraction dissolved in methanol, 5 mL of 10-fold diluted Folin-Ciocalteu solution and 4 mL of 1 mol L<sup>-1</sup> sodium carbonate were added and

incubated in the dark for 30 minutes before the absorbance was measured at 765 nm. Each sample was analysed in triplicate, with methanol as the blank. A standard curve of gallic acid was plotted and linear regression was used to estimate the gallic acid equivalence (GAE) of each extract/fraction, which was expressed as mg gallic acid per g dry weight of extract or fraction (mgGAE/gDW).

##### Total flavonoid content (TFC)

The TFC was analysed using aluminium chloride method (Ebrahimzadeh *et al.*, 2008). To each of the 0.5 mL aliquots of various concentrations of quercetin (10, 25, 50, 100, 150, 200 and 250 mg mL<sup>-1</sup>, dissolved in methanol) or 1000 µg mL<sup>-1</sup> of the extract or fraction, dissolved in methanol, 100 µL each of 10% aluminium chloride and 10% potassium acetate, 1.5 mL methanol and 2.8 mL distilled water were added. The absorbance was measured at 415 nm after a 30-minute incubation. Each sample was analysed in triplicate, with methanol as the blank. A standard curve of quercetin was plotted and linear regression was used to estimate the quercetin equivalence (QE) of each extract/fraction, which was expressed as mg quercetin per g dry weight of extract or fraction (mgQE/gDW).

##### DPPH radical scavenging activity

The DPPH radical scavenging activity of extract or fraction was evaluated following the procedure given in the literature (Vélez-González *et al.*, 2008) with slight modification. To each of

the 200  $\mu\text{L}$  aliquots of the reference standards (Quercetin or Trolox; 1, 5, 10, 25 and 50  $\mu\text{g mL}^{-1}$ , dissolved in methanol) or extracts/fractions (1, 10, 25, 50, 100, 250, 500, 750 and 1000  $\mu\text{g mL}^{-1}$ , unless otherwise stated), 1 mL of the 50  $\mu\text{g mL}^{-1}$  methanolic DPPH solution was added and incubated in the dark for 30 minutes. For control, 200  $\mu\text{L}$  of 100% methanol was used instead. Absorbance was measured at 517 nm with methanol as the blank. All samples were measured in triplicate. The hexane fraction did not show any inhibition within the range of 1 to 750  $\mu\text{g mL}^{-1}$ , therefore the assay was further carried out at higher concentrations of 1000, 1500, 2000 and 2500  $\mu\text{g mL}^{-1}$ .

The percentage DPPH radical scavenging activity was calculated using the following formula:  $[(\text{Abs}_{\text{control}} - \text{Abs}_{\text{sample}}) / (\text{Abs}_{\text{control}})] \times 100\%$ , where  $\text{Abs}_{\text{control}}$  is the measured absorbance with the methanol control, while  $\text{Abs}_{\text{sample}}$  is the measured absorbance with the reference standards or extracts. To calculate  $\text{IC}_{50}$ , which is the concentration of extract that showed 50% inhibition, radical scavenging activity against concentration was plotted and the concentration was estimated via linear regression.

#### Antibacterial analysis

The antibacterial activities of each extract were evaluated against four bacterial strains i.e. *Bacillus subtilis* [American Type Culture Collection (ATCC) cat. no. 6633], *Staphylococcus aureus* (ATCC 25923), *Escherichia coli* (ATCC 25922) and *Pseudomonas aeruginosa* (ATCC 27853). Disc diffusion assay was used as previously described (Kuate *et al.*, 2008; Zhang *et al.*, 2012; Snoussi *et al.*, 2016). All bacterial strains were grown in nutrient broth for 24 hours at 37°C before they were adjusted to 0.5 McFarland standard and inoculated onto MHA plates. A 10  $\mu\text{L}$  aliquot of extract/fraction (100, 300 and 500  $\text{mg mL}^{-1}$  in methanol) was impregnated onto a sterile 6 mm filter paper disc and then placed onto the plates. A disc with 100% methanol was used as the negative control, while a disc with 0.5  $\mu\text{L}$  of 20  $\text{mg mL}^{-1}$  streptomycin (antibiotic) as the reference standard. The plates

were then incubated at 37°C overnight before the measurement of inhibition zone was carried out. The test was done in triplicate for each extract against each microorganism. The aqueous extract was not tested due to insufficient yield obtained from extraction.

#### Statistical analysis

Statistical analysis was conducted by using one-way ANOVA with posthoc Tukey HSD test, where  $p$  value < 0.05 was considered as significant.

## RESULTS AND DISCUSSION

#### Extracts/Fractions

In total, there were two crude extracts and four partition fractions obtained in this study and the percentage yields are shown in Table 1. Soxhlet extraction with methanol produced higher yield compared to aqueous extraction, whereas partitioning with chloroform produced higher yield compared to the other solvents. This is due to their different solubility properties in the different solvents (Truong *et al.*, 2019).

#### Phytochemical screening

From Table 2, it was found that all of the extracts/fractions contained all of the five tested phytochemicals except for the hexane and diethyl ether fractions. Flavonoids and saponins were not

**Table 1.** The appearances and yields of *D. suffruticosa* leaves extracts

Extract/Fraction	Appearance	Yield (%)
Crude		
Methanol	Brown-green powder	21.8
Aqueous	Brown powder	9.2
Partition		
Hexane	Green semi-solid	5.6
Chloroform	Green powder	13.4
Ethyl acetate	Deep brown powder	5.1
Diethyl ether	Red-orange powder	1.7

**Table 2.** The phytochemical screening of *D. suffruticosa* leaves extracts/fractions

Extracts/Fractions	Alkaloids	Steroids	Phenolics	Flavonoids	Saponins
Methanol	++	+++	++	+	++
Aqueous	+	+	+	+	+
Hexane	+	+++	++	-	-
Chloroform	++	++	+	+	++
Ethyl acetate	+	++	+++	++	++
Diethyl ether	++	++	+++	++	-

+++ : indicates strong positive test; ++ : low positive test; + : weak positive test; - : negative test.

found in the hexane fraction while the diethyl ether fraction did not show the presence of saponins. Flavonoids and saponins are polar compounds, hence they would not dissolve in hexane, which is a non polar solvent. As for the absence of saponins in diethyl ether, this could be explained by the fact that partitioning with this solvent was done last. All the saponins had been extracted into the more polar chloroform and ethyl acetate prior to diethyl ether extraction. Compared with the others, both the ethyl acetate and diethyl ether fractions showed better indication in the presence of phenolics and flavonoids that are known to contribute in antioxidant activities.

The phytochemical screening of *D. suffruticosa* leaves extracts have not been previously reported. However, a previous study was carried out on the methanol extract of *D. suffruticosa* roots, which reported the presence of alkaloids, steroids, phenolics, flavonoids and saponins (Armania *et al.*, 2013). Although leaves were not used in this previous study, it is known that some compounds that are present in the roots could also be present in the leaves of the same plants but may be at different concentrations due to the polarity of methanol, where it has the ability to extract these phytochemicals either from leaves or roots (Karimi *et al.*, 2011; Senguttuvan *et al.*, 2014).

## Antioxidant activities

### Total phenolic and flavonoid content

The total phenolic content (TPC) of the extracts/fractions of *D. suffruticosa* leaves are shown in Table 3, calculated using the gallic acid standard curve ( $y = 0.004x + 0.140$ ,  $R^2 = 0.997$ ; data not shown). The methanol extract had a higher TPC of  $309.91 \pm 3.42$  mgGAE/gDW compared to the aqueous extract ( $17.86 \pm 2.71$ ) mgGAE/gDW, whereas for the fractions, the diethyl ether fraction showed the highest TPC of  $757.17 \pm 9.67$  mgGAE/gDW, followed by the ethyl acetate fraction ( $502.75 \pm 14.74$ ) mgGAE/gDW and then by the chloroform

fraction ( $64.92 \pm 1.80$ ) mgGAE/gDW, while the hexane fraction showed the lowest content among them ( $4.03 \pm 1.55$ ) mgGAE/gDW. Statistical analysis showed significant differences between all of them ( $p < 0.05$ ) except for the hexane fraction and aqueous extract, whereby these two were not significantly different from each other ( $p = 0.279$ ).

The total flavonoid content (TFC) were also shown in Table 3, calculated from the quercetin standard curve ( $y = 0.004x + 0.015$ ,  $R^2 = 0.996$ ; data not shown). As observed with the TPCs, the TFCs of both the ethyl acetate ( $89.69 \pm 7.92$  mgQE/gDW) and diethyl ether ( $100.38 \pm 7.73$  mgQE/gDW) fractions were higher compared to the others. Statistical analysis showed significant differences ( $p < 0.05$ ) between all of them, excluding between the methanol extract and hexane fraction ( $p > 0.05$ ). This could suggest that the concentrations of flavonoids between the methanol and hexane partition extracts were similar but this did not necessarily mean that they were of the same type, as there are different varieties of flavonoids known to exist in plants.

Methanol solvent has been reported to be a better medium for extraction of polyphenols (Boeing *et al.*, 2014; Asghar *et al.*, 2016;). This was observed in the methanol extract with a TPC of  $309.91 \pm 3.42$  mgGAE/gDW but the opposite was observed in the TFC of  $36.24 \pm 1.36$  mgQE/gDW, whereby the aqueous extract showed a significantly higher TFC of  $54.63 \pm 3.13$  mgQE/gDW than the methanol extract. This could be caused by the presence of other compounds that behave like flavonoids, such as amino acids, that may contribute to the TFC assay (Kolar *et al.*, 2011).

When comparing the first three consecutive fractions in Table 3, both the TPC and TFC increased with the increase in the solvent polarity of the partitioning solvent, with hexane being the less polar, followed by chloroform and ethyl acetate being the most polar. This observation was consistent with previous studies, whereby an increase in the solvent polarity increases the extraction of

**Table 3.** TPC, TFC and DPPH radical scavenging activities of *D. suffruticosa* leaf extracts/fractions and reference standards

	Total phenolic content GAE (mgGAE/gDW)	Total flavonoid content QE (mgQE/gDW)	DPPH radical scavenging IC <sub>50</sub> / $\mu\text{g mL}^{-1}$
Methanol	$309.91 \pm 3.42$	$36.24 \pm 1.36$	$305.09 \pm 4.53$
Aqueous	$17.86 \pm 2.71$	$54.63 \pm 3.13$	$1168.51 \pm 10.18$
Hexane	$4.03 \pm 1.55$	$38.77 \pm 1.64$	$2923.47 \pm 114.10$
Chloroform	$64.92 \pm 1.80$	$62.42 \pm 2.61$	$572.00 \pm 10.24$
Ethyl acetate	$502.75 \pm 14.74$	$89.69 \pm 7.92$	$29.42 \pm 0.49$
Diethyl ether	$757.17 \pm 9.67$	$100.38 \pm 7.73$	$84.60 \pm 2.74$
Quercetin	–	–	$16.36 \pm 0.66$
Trolox	–	–	$38.04 \pm 0.53$

Values shown are average  $\pm$  SD of triplicate; GAE: gallic acid equivalence; QE: quercetin equivalence; DW: dry weight of extract/fraction.



active polar compounds (Goze *et al.*, 2009; Vats, 2012; Addai *et al.*, 2013; Belyagoubi *et al.*, 2016). However, as seen in Table 3, the diethyl ether fraction interestingly showed the highest TPC of  $757.17 \pm 9.67$  mgGAE/gDW and TFC of  $100.38 \pm 7.73$  mgQE/gDW compared to the other fractions, despite being less polar than chloroform and ethyl acetate. It should be noted that the fractionation with diethyl ether was the last step of the solvent-solvent partitioning, after the formation of yellow precipitate was observed, as described in the methodology. The compounds of this precipitate could have slowly accumulated as other compounds were extracted consecutively with hexane, chloroform and ethyl acetate, and consequently they became increasingly insoluble and eventually precipitated (Kebbab-Massime *et al.*, 2017). Hence, it is possible that a less polar solvent could extract more polyphenols than the more polar ones via precipitation.

#### **DPPH radical scavenging activity**

The DPPH radical scavenging activities, expressed as IC<sub>50</sub>, of the extracts/fractions and two reference standards, quercetin and Trolox, are also shown in Table 3. The IC<sub>50</sub> values were significantly different from each other ( $p < 0.05$ ) except for the ethyl acetate and diethyl ether fractions. When compared to the standards, it was observed that ethyl acetate fraction showed an IC<sub>50</sub> activity of  $29.42 \pm 0.49$   $\mu\text{g mL}^{-1}$  that was comparable to Quercetin ( $16.36 \pm 0.66$   $\mu\text{g mL}^{-1}$ ) and Trolox ( $38.04 \pm 0.53$   $\mu\text{g mL}^{-1}$ ). In addition, this is consistent with the findings from Armania *et al.* (2013) where an IC<sub>50</sub> value of  $31.33 \pm 1.15$   $\mu\text{g mL}^{-1}$  was obtained for the methanolic extracts from the roots of *D. suffruticosa*. It is known that some compounds that are present in the roots could also be present in the leaves of the same plants as mentioned earlier (Karimi *et al.*, 2011; Senguttuvan *et al.*, 2014). Therefore, the ethyl acetate fraction showed the strongest radical scavenging activity. This is as expected when taking into account of its high TPC and TFC values. In contrast, the hexane fraction showed the least radical scavenging activity with IC<sub>50</sub> value of  $2923.47 \pm 114.10$   $\mu\text{g mL}^{-1}$  and the low TPC of  $4.03 \pm 1.55$  mgGAE/gDW and TFC of  $38.77 \pm 1.64$  mgQE/gDW observed in this extract could explain this low activity.

It was also observed that the DPPH radical scavenging activity also increased with the increase in the polarity of the extracting solvent, as observed in the TPC and TFC assays. Moreover, the methanol extract ( $305.09 \pm 4.53$   $\mu\text{g mL}^{-1}$ ) also exhibited a stronger antioxidant activity compared to the aqueous extract ( $1168.51 \pm 10.18$   $\mu\text{g mL}^{-1}$ ). This could be because, as reported in a previous study (Sulaiman *et al.*, 2011), water has low efficiency in

extracting antioxidant compounds from natural products, and it was shown that most of the tested aqueous plant extracts were least effective in scavenging the DPPH radicals compared to the organic-aqueous extracts.

There is only one previous study on the antioxidant activities of the crude methanol extract of *D. suffruticosa* leaves, which was obtained by maceration of the powdered leaves for 72 hours (Armania *et al.*, 2013), however, in our study, the crude methanol extract was obtained by Soxhlet extraction. The TPC of  $309.91 \pm 3.42$  mgGAE/gDW observed in our study is different from the TPC of  $236.49 \pm 2.37$  mgGAE/gDW reported previously (Armania *et al.*, 2013) suggests that Soxhlet extraction is better in extracting phenolic compounds than by maceration. This is supported by a previous study that evaluated these different extraction methods and found that Soxhlet extraction ( $133.70$  mg GAE/ g extract) had extracted more phenolic compounds than by maceration ( $101.31$  mg GAE/ g extract) (Sharma & Cannoo, 2016).

A previous study on the leaves of *D. indica*, which is a close relative of *D. suffruticosa*, reported that it inhibited the DPPH radicals by 82.32% using  $500$   $\mu\text{g mL}^{-1}$  of the extract (Kumar *et al.*, 2011). At this same concentration, the crude methanol extract of *D. suffruticosa* in this present study similarly inhibited the DPPH radicals by 79.93%. Despite being two different species, similar DPPH radical scavenging activities were observed. Hence, although *D. indica* is more frequently studied for its medicinal purposes, *D. suffruticosa* also has similar potential as *D. indica* and should not be ignored but warrant further studies. In addition, it was reported in another previous study that the high antioxidant activities seen in the *D. indica* crude extracts could be due to the synergistic activity of all the active components present in the extracts, not just from one particular compound (Rashid *et al.*, 2009).

#### **Antibacterial activities**

Plants have been known to produce various chemical components that have different biological activities against various microorganisms (Matić *et al.*, 2016). For the current study, the antibacterial activity was detected when a clear or semi-clear inhibition around the disc containing the extract was present (Alam *et al.*, 2011). The zones of inhibition for the *D. suffruticosa* extracts, as well as for the reference standard, against four bacterial species, are shown in Table 4. The negative control (100% methanol) did not show any zone of inhibition to any of the bacterial species, as expected. Generally, the results showed that under the conditions tested, no detectable antibacterial activity was observed against the Gram-negative bacteria, *E. coli* and *P.*

**Table 4.** Zone of inhibition of *D. suffruticosa* methanol extract and its fractions

Extract	Microorganism	Zone of inhibition (mm)			
		Streptomycin	100 mg mL <sup>-1</sup>	300 mg mL <sup>-1</sup>	500 mg mL <sup>-1</sup>
<b>Methanol</b>	<i>S. aureus</i>	13.11 ± 1.76	7.17 ± 1.22	7.44 ± 0.98	6.89 ± 0.78
	<i>B. subtilis</i>	23.44 ± 3.4	–	–	–
	<i>E. coli</i>	21.33 ± 0.71	–	–	–
	<i>P. aeruginosa</i>	11.67 ± 1.15	–	–	–
<b>Hexane</b>	<i>S. aureus</i>	15.11 ± 0.601	11.67 ± 2.00	10.11 ± 2.26	7.72 ± 0.75
	<i>B. subtilis</i>	24.89 ± 1.45	6.33 ± 0.50	6.39 ± 0.49	6.78 ± 0.51
	<i>E. coli</i>	21.33 ± 0.58	–	–	–
	<i>P. aeruginosa</i>	12.00 ± 1.00	–	–	–
<b>Chloroform</b>	<i>S. aureus</i>	14.44 ± 1.13	6.56 ± 0.46	6.83 ± 0.56	6.72 ± 0.83
	<i>B. subtilis</i>	23.67 ± 1.58	6.39 ± 0.42	6.61 ± 0.60	6.94 ± 0.81
	<i>E. coli</i>	21.67 ± 0.58	–	–	–
	<i>P. aeruginosa</i>	12.22 ± 0.97	–	–	–
<b>Ethyl acetate</b>	<i>S. aureus</i>	14.89 ± 0.78	6.83 ± 0.35	8.11 ± 0.78	8.78 ± 0.97
	<i>B. subtilis</i>	24.78 ± 2.44	–	–	–
	<i>E. coli</i>	21.00 ± 1.00	–	–	–
	<i>P. aeruginosa</i>	12.33 ± 1.15	–	–	–
<b>Diethyl ether</b>	<i>S. aureus</i>	14.33 ± 0.71	10.56 ± 2.60	12.06 ± 2.77	12.22 ± 1.66
	<i>B. subtilis</i>	24.89 ± 1.27	–	6.50 ± 0.50	7.28 ± 0.97
	<i>E. coli</i>	21.33 ± 0.58	–	–	–
	<i>P. aeruginosa</i>	12.67 ± 0.58	–	–	–

Values shown are average ± SD of triplicate; – means no growth inhibition zone.

*aeruginosa*. However, antibacterial activities were observed against the Gram-positive bacteria, *S. aureus* and *B. subtilis*.

From Table 4, it is shown that the antibacterial activity of the crude methanol extract was only detected against *S. aureus* but not the other bacteria. In contrast, a previous antimicrobial study on the same plant species (Wiarat *et al.*, 2004) showed that a 1 mg methanol extract of *D. suffruticosa* had antibacterial activities against *B. subtilis*, with zone of inhibition of only 7 mm and *P. aeruginosa*, with inhibition zone of 9 mm. It also had detectable activities against *B. cereus* and *Candida albicans*. This could perhaps be due to variations in the protocols or strains used. Until now, there is no previous antimicrobial study with regards to the solvent-solvent partition extracts of *D. suffruticosa*. However, there is a previous antimicrobial study of its close relative, *D. indica*, which did solvent-solvent partitioning using hexane, carbon tetrachloride, dichloromethane and chloroform. This previous study reported that only the chloroform fraction exhibited antimicrobial activity (Rashid *et al.*, 2009). In contrast, in the present study antibacterial activities were detected in all of the fractions. This might be due to *D. indica* being a different species to that of *D. suffruticosa*, and therefore would have different phytochemical contents that lead to the results observed here.

Antibacterial activity against *B. subtilis* was only observed with the hexane, chloroform and diethyl ether fractions (Table 4). The results suggest

that the hexane and chloroform fractions were more potent than the diethyl ether fraction in inhibiting *B. subtilis*, as inhibition zone could still be detected at 100 mg mL<sup>-1</sup> of either the hexane (6.33 ± 0.50 mm) or chloroform fraction (6.39 ± 0.42 mm) but not with the diethyl ether fraction. The methanol extract and its fractions showed antibacterial activity against *S. aureus* (Table 4). However, based on the diameter of the inhibition zone, it seemed that *S. aureus* was mostly affected by the diethyl ether fraction (10.56 ± 2.60 to 12.22 ± 1.66 mm) compared to the others. Hence, it is possible that the antibacterial compounds in this plant species might have more affinity towards diethyl ether. We observed a slight decrease in the zone of inhibition for both methanol extract and hexane fraction for *S. aureus*, however, this could be due to solubility decreasing at higher concentrations. The higher TPC and TFC of the diethyl ether fraction might explain the potency of this fraction against *S. aureus*. The polar polyphenolic compounds in the diethyl ether fraction was concentrated and precipitated after the successive partitioning (Kebbab-Massime *et al.*, 2017) despite the solvent being less polar. It has been previously reported that polyphenolic compounds play a substantial role in the antioxidant and antimicrobial activities of plant extracts (Aker *et al.*, 2016; Benabdelaziz *et al.*, 2016; Habibatni *et al.*, 2016; Olivier *et al.*, 2017). Since phenolic compounds or flavonoids have variations in their structures, different compounds

may exhibit different antioxidant and antimicrobial activities. These phenolics and flavonoids can be synthesised by plants as a defence mechanism against microbial infection (Akter *et al.*, 2016; Habibatni *et al.*, 2016).

There is no obvious pattern observed between the antibacterial activity and solvent polarity i.e. no increase in activity with increasing polarity of the first three solvents (hexane, chloroform and ethyl acetate) as seen earlier in the TPC, TFC and DPPH radical scavenging assays. This seems to suggest that the polarity of solvent did not play an important role in the extraction of antibacterial compounds. This could also suggest that there is no strong correlation between the antibacterial activity and antioxidant activity i.e. the high antioxidant activity did not correspond to high antibacterial activity. A similar observation was previously reported whereby the plant extracts, which had shown high antioxidant activities, showed low inhibition of bacterial growth (Borchardt *et al.*, 2008; Koncic *et al.*, 2010; Jang *et al.*, 2016; Vitalini *et al.*, 2016).

## CONCLUSION

This is the first study on the antioxidant and antibacterial activities of the *D. suffruticosa* extracts obtained using decoction, Soxhlet extraction and solvent-solvent partitioning. In summary, the study showed that *D. suffruticosa* leaves had antioxidant and antibacterial activities, with the diethyl ether and ethyl acetate partition extracts being the most promising extracts. The discovery of new pharmaceutical or effective antimicrobial agents could be explored by further investigation on the bioassay-guided isolation of bioactive compounds of *D. suffruticosa*.

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